PRE AP CHEMISTRY			
Mrs. Suzanne van Balveren			
	suzannevanbalveren@laschina.org		
http://sciencelas.weebly.com			
Room 318			
Office hours: Monday - Friday 8.00 - 8.30, 3.45 – 4.15			
Tutoring: Mondays 3.45 - 4.45			
<u>Mission:</u>	to be critical and independent thi effectively in a scientific and tech	nnological society. Students will be cietal issues using scientific problem	
Textbook and materials:	 Textbook - Modern Chemistry by Sarquis & Sarquis, ISBN: 9780547586632 A variety of materials gathered from other sources 		
	It is important that students come to class prepared every day! Unprepared students will have a hard time keeping up, and therefore will lose participation points for unpreparedness.		
	Materials: Journal Three Ring Binder with 5 Divider Calculator Pencil and Eraser Pen Ruler Highlighters VPN; teacher will show videos al will also provide extra study mate	bout various concepts in class, but	
<u>Grades:</u>	$\begin{array}{l} 90\% - 100\% &= A\\ 80\% - 89\% &= B\\ 70\% - 79\% &= C\\ 65\% - 69\% &= D\\ 64\% - below &= F\\ Incomplete &= I\\ \hline \\ \hline \\ \hline \\ Course grading:\\ \hline \\ Tests\\ \hline \\ Projects, Labs, Lab reports, etc.\\ \hline \\ Quizzes\\ \hline \\ Class work and Participation\\ \hline \\ Homework\\ \hline \end{array}$	40% 20% 20% 10%	
	ΠΟΠΙΕΨΟΙΚ	1070	

Absent work and Late Work Policy	If a student is absent, the amount of time allotted to complete missing assignments is equal to the amount of school missed. If a student is absent two days, he/she will have two days to complete the missing work.	
	This course will cover an insane amount of content at a fast pace, I advise students to be present every day. Students planning to miss multiple days of school should contact me <u>prior</u> to departure to discuss a study plan.	
	It is the student's responsibility to meet with classmates to collect missed lecture notes.	
	If a student misses a quiz or test, they should meet with me directly to reschedule the assessment.	
	Late work: In order to not fall behind, I expect students to complete all assignments on time. Late assignments will lose ten percent for each (school) day late.	
Website:	http://sciencelas.weebly.com	
	This website is currently under construction and will be completed in September. Once the website is up and running students will be able to access course information online. For example: PPT Slides, videos, review sheets etc.	
Expectations:		

Plagiarism	 4. Confident Individuals: Take responsibility over yourself and your work. Take pride in what you do and always do your best work. Motivate yourself to take an authoritative role in your learning. Plagiarism and cheating will not be tolerated. Any dishonest 	
and cheating:	behavior of this nature will result in a failing grade on the assignment and a meeting with administration and parents.	
<u>Tardy policy:</u>	Students are expected to be in the seats working on the "DO NOW" when the bell rings.Tardy policy is according to "secondary handbook 2015-2016": 1 st tardy: Warning 2 nd tardy: Referral to After-School Detention 3 rd tardy: Referral to After-School Detention Teacher contacts parents4 th tardy: Count as unexcused absence Referral to After-School Detention 5 th tardy: One day In-School Suspension Meeting with Parents Administrator Discretion	
Cell phones and other electronics:	No cellphones are allowed in the classroom. If you bring your cellphone to class it will be confiscated. You will be able to pick up your phone after school. Other electronic devices may only be used <u>after</u> approval of the teacher.	
<u>Content</u> <u>Outline:</u>	Quarter 1:Lab safety and lab reportsChapter 1:Matter and changeChapter 1:Matter and changeChapter 2:Measurements and calculationsChapter 3:Atoms: The building blocks of matterChapter 4:Arrangements of electrons in atomsChapter 5:The periodic lawChapter 6:Chemical Bonding:Chapter 7:Chemical formulas and chemical compoundsQuarter 2:Chapter 8:Chemical equations and reactionsChapter 9:StoichiometryChapter 10:States of matterChapter 11:GasesChapter 12:Solutions	

Chapter 14 Chapter 15 Chapter 10	 B: Ions in aqueous solutions 4: Acids and bases 5: Acid-base titrations and pH 6: Reaction energy 7: Reaction kinetics
Chapter 19 Chapter 20	 3: Chemical equilibrium 3: Oxidation-Reduction reactions 5: Electrochemistry 1: Nuclear Chemistry